<u>Review 02</u>

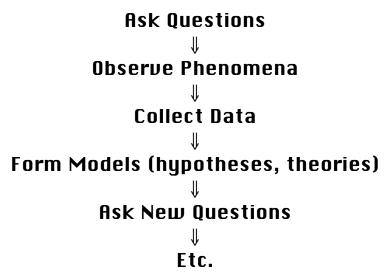
What is science?

- •a systematic search for an answer
- •a system of <u>asking questions</u> in such a way that <u>clear-</u> <u>cut answers</u> will surface
- a practice of observing phenomena in such a manner that <u>unexpected results</u> will be <u>acceptable</u>
- •a way of accumulating facts and <u>forming models which can be</u> <u>further tested</u> to give light to new facts
- •a <u>methodology</u> used to come to understand the natural phenomena which surrounds us and permeates our lives

Bottom line: <u>a way of thinking and doing</u>

Don't confuse science with technology! Basic science develops principles, which underlie technology. Technology is making use of these principles.

How we think and do:



Notes on experiments: ONE variable at a time! i.e., make only one change at a time Experimental control! unmodified sample against which to compare modified sample Science is very systematic!

Review 02, con't

Biology--the scientific study of life

Characteristics of living organisms

Levels of organization of life

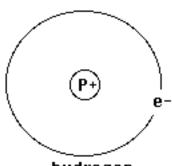
Organization of non-living world into living matter

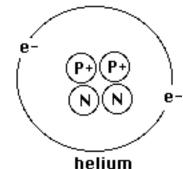
Carbon, Hydrogen, Oxygen, Nitrogen, Phosphorus, Sulfur

✓ these 6 elements which make up 99% of living matter

>>> SPONCH <<<

- \checkmark they can be arranged in various ways to make
 - water
 - proteins (amino acids)
 - lipids (fats & oils)
 - carbohydrates (sugars)
 - nucleic acids
- Elements are composed of atoms
- Atoms are composed of electrons, protons and neutrons
- Elements are identified by number of protons in the atomic nucleus





hydrogen

nellum